# Simultaneous CT-<sup>13</sup>C and VT-<sup>15</sup>N chemical shift labelling: Application to 3D NOESY-CH<sub>3</sub>NH and 3D <sup>13</sup>C, <sup>15</sup>N HSQC-NOESY-CH<sub>3</sub>NH

Dušan Uhrín<sup>a,\*</sup>, Janice Bramham<sup>a,b</sup>, Steven J. Winder<sup>b</sup> & Paul N. Barlow<sup>a</sup>

<sup>a</sup>University of Edinburgh, Edinburgh Centre for Protein Technology, Joseph Black Chemistry Building, West Mains Rd., Edinburgh EH9 3JJ, Scotland; <sup>b</sup> University of Glasgow, Institute of Biomedical and Life Sciences, Division of Biochemistry and Molecular Biology, Davidson Building, Glasgow G12 8QQ, Scotland

Received 13 July 2000; Accepted 30 August 2000

Key words: constant and variable time chemical shift labelling, HSQC, 3D <sup>13</sup>C, <sup>15</sup>N HSQC-NOESY-CH<sub>3</sub>NH

#### Abstract

Based on the HSQC scheme, we have designed a 2D heterocorrelated experiment which combines constant time (CT)  ${}^{13}$ C and variable time (VT)  ${}^{15}$ N chemical shift labelling. Although applicable to all carbons, this mode is particularly suitable for simultaneous recording of methyl-carbon and nitrogen chemical shifts at high digital resolution. The methyl carbon magnetisation is in the transverse plane during the whole CT period (1/J<sub>CC</sub> = 28.6 ms). The magnetisation originating from NH protons is initially stored in the 2H<sub>z</sub>N<sub>z</sub> state, then prior to the VT chemical shift labelling period is converted into 2H<sub>z</sub>N<sub>y</sub> coherence. The VT  ${}^{-15}$ N mode eliminates the effect of  ${}^{1}J_{N,CO}$  and  ${}^{1.2}J_{N,CA}$  coupling constants without the need for band-selective carbon pulses. An optional editing procedure is incorporated which eliminates signals from CH<sub>2</sub> groups, thus removing any potential overlap with the CH<sub>3</sub> signals. The CT- ${}^{13}$ CH<sub>3</sub>,VT- ${}^{15}$ N HSQC building block is used to construct two 3D experiments: 3D NOESY-CH<sub>3</sub>NH and 3D  ${}^{13}$ C, ${}^{15}$ N HSQC-NOESY-CH<sub>3</sub>NH. Combined use of these experiments yields proton and heteronuclear chemical shifts for moieties experiencing NOEs with CH<sub>3</sub> and NH protons. These NOE interactions are resolved as a consequence of the high digital resolution in the carbon and nitrogen chemical shifts of CH<sub>3</sub> and NH groups, respectively. The techniques are illustrated using a double labelled sample of the CH domain from calponin.

#### Introduction

Evaluation of interproton distances is the essential step in determination of protein structures by NMR spectroscopy. For <sup>13</sup>C and <sup>15</sup>N double-labelled proteins, 4D NOESY experiments with two proton and two heteronuclear chemical shift axes provide, in principle, all the data required for identification of protons located close in space (Kay et al., 1990; Clore et al., 1991; Zuiderweg et al., 1991). However, the limited digital resolution and sensitivity inherent to 4D experiments mean that, in practice, several 3D NOESY experiments usually have to be recorded instead (Bax and Grzesiek, 1993). Nonetheless, the fundamental limitations in the digital resolution of the <sup>13</sup>C dimension in 3D <sup>13</sup>C-edited NOESY spectra together with degeneracy of chemical shifts make the assignment of 3D NOESY spectra of proteins one of the most challenging problems in structure determination by NMR.

Several 3D experiments have shown that for methyl groups, whose carbon spin–spin relaxation times are generally longer than those of the rest of the amino acid side chains, the digital resolution in the <sup>13</sup>C dimension can be increased substantially without dramatic decrease of the cross peak intensities (Hu and Zuiderweg, 1996; Diercks et al., 1998; Zwahlen et al., 1998a, b; Uhrín et al., 2000). High digital resolution NOE data for methyl protons increase the accuracy of NMR structures as methyl-containing amino acids occur frequently in hydrophobic cores of proteins

<sup>\*</sup>To whom correspondence should be addressed. E-mail: du-san@chem.ed.ac.uk



Figure 1. (a) Pulse sequence of CT-<sup>13</sup>CH<sub>3</sub>, VT.<sup>15</sup>N HSQC. Thin and thick bars represent 90° and 180° non-selective pulses and are applied along the x-axis unless otherwise stated. The delays are:  $\tau_1 = 2.25 \text{ ms}$ ,  $\tau_2 + \text{pw}_{sel} = \tau_1$ , where  $\text{pw}_{sel}$  is a 90° <sup>1</sup>H selective pulse, T = 28.6 ms,  $\delta_1 = (T - t_2)/2 - \text{pwn} + (4/\pi)\text{pwn} + \text{pwc}$ ,  $\delta_4 = t_2/2 - (2/\pi)$  pwn,  $\delta_2 = \delta_4 - \text{pwh}$ ,  $\delta_3 = \delta_4 - \text{pwh} - \text{pwc}$ ,  $\delta_5 = (T - t_2)/2 - \text{pwn} + (4/\pi) \text{pwn} - \tau_a$ ,  $\delta_6 = \tau_a - \text{pwn}$ , where  $\tau_a = 4.0 \text{ ms}$  and pwn, pwc and pwh are 90° <sup>15</sup>N, <sup>13</sup>C and <sup>1</sup>H pulses, respectively. The following phases were applied:  $\varphi_1 = x, -x, \varphi_2 = x$  and  $\psi = y, -y, -y, y$ ; every two scans the central part of the sequence was replaced by the one shown in the inset. The <sup>13</sup>C and <sup>15</sup>N carriers were set to 20 and 118 ppm, respectively. The <sup>13</sup>C inversion pulse was a 692 µs I-SNOB-5 pulse (Kupče, personal communication). The sign discrimination in F<sub>1</sub> was achieved by the States-TPPI method with a 90° incrementation of the phases  $\varphi_1$  and  $\varphi_2$ . GARP1 decoupling covering the CH<sub>3</sub> and NH regions ( $\gamma B_2/2\pi = 2000 \text{ Hz}$ ) was used. (b+a) 3D NOESY-CH<sub>3</sub>NH,  $\varphi_3 = 4x, 4(-x), \psi = y, -y, -y, y, -y;$  the carbon decoupling during t<sub>1</sub> was achieved by a 126 µs broadband inversion pulse (Smith and Shaka, 2000). The sign discrimination in F<sub>1</sub> was achieved by the States-TPPI method with equivalence of the sequence use of  $1^3 \text{ C}, 1^5 \text{ N}$  HSQC-NOESY-CH<sub>3</sub>NH,  $\tau_3 = 2.6 \text{ ms}, \delta_7 = t_1/2 - pwc - (2/\pi) (pwn - pwc), \delta_8 = t_1 - (4/\pi) pwc$ . The adiabatic inversion WURST-20 pulse applied after the 180° <sup>1</sup>H pulse was optimised to cover only the aliphatic region. The field was swept through 27 kHz during 1.098 ms from higher to lower field with the carrier positioned at 30 ppm (Kupče and Freeman, 1997). Phase  $\varphi_3$  and  $\psi$  were identical to those used in (b), phase  $\varphi_4 = x$  was changed to -x after eight scans and the data were stored into a separate memory location.

where they have an intrinsic structural role involving many inter-residue interactions. Methyl groups are also represented on protein surfaces where they can be involved in intermolecular interactions (Gardner et al., 1997). NOESY experiments focusing on methyl groups (Diercks et al., 1998; Zwahlen et al., 1998a, b) can also be designed to incorporate detection of the NOE connectivity of NH protons (Zwahlen et al., 1998b). Such a combination is a part of a continued effort to reduce the number of necessary 3D NOESY experiments by maximising the information content of each experiment (Sørensen, 1990; Pascal et al., 1994; Vis et al., 1994; Jarela and Rule, 1995; Zwahlen et al., 1998b).

We discuss here two 3D <sup>13</sup>C,<sup>15</sup>N NOESY experiments designed to yield interproton NOEs between the following pairs of groups; CH<sub>3</sub> and CH<sub>n</sub>, NH and CH<sub>n</sub>, and NH and NH (n = 1,2,3). The spectra are acquired with high digital resolution in the combined <sup>13</sup>CH<sub>3</sub> and <sup>15</sup>N F<sub>2</sub> dimension. The directly detected dimension contains CH<sub>3</sub> and NH protons only, while the F<sub>1</sub> dimension is used for labelling of either <sup>1</sup>H or <sup>13</sup>C and <sup>15</sup>N chemical shifts. The experiments are illustrated using a double labelled sample (12.5 kDa, 1.2 m M, pH 7.0, 18 °C ) of the CH domain from calponin (I).

#### **Results and discussion**

## Combined CT-<sup>13</sup>CH<sub>3</sub>, VT-<sup>15</sup>N HSQC

At the heart of the proposed NOESY experiments is a combination of a constant (CT) and variable (VT)



*Figure 2.* (a)  $CT^{-13}CH_3$ ,  $VT^{-15}N$  HSQC spectrum of **I** acquired using the pulse sequence of Figure 1a. The sweep width in F<sub>1</sub> was 2600 Hz, 4 scans were accumulated into each of 54 complex points. The constant time interval, T, was set to 28.6 ms. Acquisition times were 20.8 and 64 ms in t<sub>1</sub> and t<sub>2</sub>, respectively and the total acquisition time was 10 min. (b)  $CT^{-15}N$  HSQC spectrum acquired using identical parameters to those in (a) using the constant-time chemical shift labelling scheme as shown in (c). The nitrogen–carbon interactions were decoupled by a 126 µs  $^{13}C$  broadband inversion pulse (Smith and Shaka, 2000).

time <sup>13</sup>CH<sub>3</sub> and <sup>15</sup>N-HSQC (Figure 1a). This combination of labelling periods has, to our knowledge, not been described previously. The magnetisation of the NH protons is initially stored in the form of two-spin order  $(2H_zN_z)$  and converted into nitrogen antiphase magnetisation  $(2H_zN_y)$  only during the later part of the CT interval. The length of the CT CH<sub>3</sub> period  $(1/{}^1J_{CC})$ is dictated by the need to refocus the one-bond carbon– carbon coupling constants. The amount of time for which the two-spin order exists is gradually reduced as the length of the VT-<sup>15</sup>N chemical shift labelling period increases. This period is centred around the <sup>1</sup>H 180° pulse which serves to refocus the <sup>1</sup>J<sub>NH</sub> (and also <sup>1</sup>J<sub>CH3</sub>) coupling constants. Elimination of the evolution of <sup>1</sup>J<sub>N,CO</sub> and <sup>1,2</sup>J<sub>N,CA</sub> is not required, since these are not resolved in  $F_1$  during the typical acquisition time of around 20 ms.

For medium size proteins the relaxation of the twospin order,  $2H_zN_z$ , is similar to that of the <sup>15</sup>N transverse magnetisation (Dayie and Wagner, 1994). The <sup>15</sup>N interferogram of the VT-<sup>15</sup>N t<sub>1</sub> period obtained in the CT-<sup>13</sup>CH<sub>3</sub>, VT-<sup>15</sup>N HSQC therefore resembles that of a CT experiment. The NH part of the spectrum of **I** acquired using the pulse sequence of Figure 1a, and the spectrum acquired using the CT-<sup>15</sup>N t<sub>1</sub> period of Figure 2c are shown for comparison in Figures 2a and b, respectively. Horizontal projections of these spectra are almost identical; the slight broadening of the resonances in F<sub>1</sub> seen in the former spectrum is due to unresolved nitrogen–carbon coupling constants.



*Figure 3.* The  $F_1F_3$  plane from a 3D NOESY-CH<sub>3</sub>NH spectrum of **I** acquired using the pulse sequence of Figure 1(b+a). The NOE mixing time was 100 ms; the acquisition times  $t_1$ ,  $t_2$  and  $t_3$  were 15, 19.2 and 64 ms respectively, corresponding sweep widths were 6400, 2500 and 8000 Hz and 96, 48 and 512 complex points were accumulated during 63 h using eight scans per increment.

A broad-band inversion  ${}^{13}$ C pulse, which was used in the CT- ${}^{15}$ N experiment (Figure 2c), is not applicable in experiments with combined CT- ${}^{13}$ CH<sub>3</sub>, ${}^{15}$ N periods (Zwahlen et al., 1998b). Selective 180° C', C $\alpha$  and CH<sub>3</sub> pulses must be used with a Bloch–Siegert shift compensation for at least the C $\alpha$  pulse. This means that a total of four selective  ${}^{13}$ C pulses are required in an experiment with a shared CT- ${}^{15}$ N, ${}^{13}$ C period (Zwahlen et al., 1998b), while only one non-selective 180°  ${}^{13}$ C pulse is needed in the proposed combination of the CT- ${}^{13}$ C and VT- ${}^{15}$ N chemical shift labelling.

The magnetisations of CH<sub>2</sub> and CH<sub>3</sub> differ in sign at the end of the constant time interval  $(1/{}^{1}J_{CC})$  due to the different number of interacting carbon nuclei. In the case of overlap, the destructive interference between these two types of signals can decrease the quality of NOESY spectra. The CH<sub>2</sub> signals can be removed from the CT- $^{13}$ CH<sub>3</sub>, VT- $^{15}$ N HSQC spectra in a way similar to that used in the 3D HCCH<sub>3</sub>-TOCSY experiment (Uhrín et al., 2000): the 180°  $^{1}$ H pulse is shifted by 4 ms (=  $1/2^{1}J_{CH3}$ ) towards the beginning of the pulse sequence during half of the scans accumulated for each increment. This causes a change in the sign of the CH and CH<sub>3</sub> resonances, but does not affect the CH<sub>2</sub> signals, which are then eliminated, by a suitable phase cycling of the receiver. A part of the CH<sub>2</sub> signal which was not completely eliminated due to the differences in  ${}^{1}J_{CH}$  between the methyl and methylene groups, is substantially reduced by the fast spin-spin relaxation of CH<sub>2</sub> carbons. The CH signals from Ca carbons are not excited at all due to the methyl selective inversion used during the de- and refocusing INEPT periods. The only signals, except for those from CH<sub>3</sub> carbons, which pass this filter, albeit much attenuated due to faster spin-spin relaxation, are those from CH groups resonating in the methyl region. Since their phase is identical to that of the CH<sub>3</sub> signals, they do not cause signal cancellation. By employing this procedure for selection of methyl resonances rather than coherence selection by pulsed field gradients, the sensitivity of the proposed method is higher than the sensitivity of the CT-QQF-HSQC experiment (Diercks et al., 1998).

#### 3D NOESY-CH<sub>3</sub>NH

A 3D NOESY-CH<sub>3</sub>NH experiment is obtained by placing the proton chemical shift labelling followed by the NOE mixing time before the  $CT^{-13}CH_3$ , VT-



*Figure 4.* Parts of the first  $F_1F_3$  plane from a 3D  ${}^{13}C, {}^{15}N$  HSQC-NOESY-CH<sub>3</sub>NH spectrum of I containing NH and CH<sub>3</sub> correlations acquired using the pulse sequence of Figure 1(c+a). (a) Contains both CH and NH cross peaks, while the spectra in (b) and (c) were obtained by adding or subtracting, respectively, the data sets acquired by changing the phase  $\varphi_4$  and stored in separate memory blocks. In this manner the CH and NH cross peaks were separated. Only one half of the data were used to give the spectrum in (a). The NOE mixing time was 100 ms. The  ${}^{13}C$  carrier was set to 30, 45 and 20 ppm for adiabatic inversion and  ${}^{13}C$  chemical shift labelling during  $t_1$  and  $t_2$ , respectively. Only the NOEs from aliphatic carbons were recorded using a  ${}^{13}C$  selective inversion pulse described in the legend of Figure 1c. The acquisition times  $t_1$  and  $t_3$  were 6.4 and 64 ms respectively, corresponding sweep widths were 9955 and 8000 Hz; 64 and 512 complex points were acquired; 16 scans were accumulated per increment using  $\varphi_4 = x$  and another 16 scans using  $\varphi_4 = -x$ . The total acquisition time was 1.5 h.

<sup>15</sup>N HSQC block (Figure 1(b+a)). A 2D plane from the 3D NOESY-CH<sub>3</sub>NH spectrum of **I** is shown in Figure 3. Four different types of cross peaks are visible here: CH<sub>3</sub>-CH<sub>n</sub>, CH<sub>3</sub>-NH, NH-NH and NH-CH<sub>n</sub> (n = 1,2,3), where CH<sub>n</sub> includes both aliphatic and aromatic protons. Superior digital resolution in the <sup>15</sup>N,<sup>13</sup>C-shared  $F_2$  dimension is a consequence of a ca. 20 ms acquisition time, which is considerably longer than the 5–8 ms typically used for <sup>13</sup>C-edited spectra. The same acquisition time applies to the <sup>15</sup>N chemical shift labelling. This is comparable to acquisition times used in the 3D <sup>15</sup>N-edited NOESY exper-

iments, but much longer than the acquisition achievable in the shared  ${}^{13}C$ ,  ${}^{15}N$ -edited NOESY experiment with simultaneous carbon and nitrogen pulses (Pascal et al., 1994). Unfortunately, there is an initial loss of magnetisation caused by the relaxation of the  $2N_zH_z$  two-spin order during the 28.6 ms period when compared with the standard 3D  ${}^{15}N$ -edited NOESY, causing some signal attenuation.

# 3D<sup>13</sup>C,<sup>15</sup>N HSQC-NOESY-CH<sub>3</sub>NH

Even when there is no overlap between NOE cross peaks in 3D <sup>13</sup>C NOESY-HSQC spectra, their assignment is not unambiguous. This is mostly due to our lack of knowledge of the chemical shift of the heteronucleus to which one of the pair of interacting protons is attached. This information can be provided by searching for symmetry related cross peaks, a procedure which is time consuming and not always successful. The missing heteronuclear chemical shift information can alternatively be obtained in a complementary 3D experiment, which employs the initial chemical shift labelling of <sup>13</sup>C or <sup>15</sup>N rather than that of protons (Frenkiel, 1990; Vis et al., 1994; Jarela and Rule, 1995). There, by working simultaneously with two 3D NOESY spectra differing only by the nucleus labelled in the  $F_1$  dimension (<sup>1</sup>H or X), and comparing the chemical shifts of cross peaks with the resonance assignment table, it is usually possible to identify unambiguously the interacting proton pair. An additional advantage of <sup>13</sup>C-edited NOESY spectra with two <sup>13</sup>C axes comes from the larger dispersion of <sup>13</sup>C chemical shifts which may uncover more NOE contacts than had been assumed previously based on the analysis of a standard <sup>13</sup>C-edited NOESY spectrum (Diercks et al., 1999).

It is also possible for <sup>15</sup>N and <sup>13</sup>C to share the chemical shift labelling period prior to the NOE mixing time. In this case, however, the problem of overlap of <sup>15</sup>N and <sup>13</sup>C resonances in F<sub>1</sub> must be addressed. The first solution, which involves elimination of the diagonal peaks by a suitable phase cycling, also results in the elimination of all NH-NH and CH-CH cross peaks (Jarela and Rule, 1995). The second, and the preferred solution, relies on the independent phase cycling of the 90° <sup>15</sup>N and <sup>13</sup>C pulses of the first IN-EPT transfer (Sørensen, 1990). The latter solution was adopted for the proposed 3D 13C,15N HSQC-NOESY- $CH_3NH$  experiment (Figure 1(c+a)). In this way, the separation of CH and NH cross peaks in F<sub>1</sub> is achieved by spectral editing, as exemplified in Figure 4 using the first plane of the 3D <sup>13</sup>C,<sup>15</sup>N HSQC-NOESY-

CH<sub>3</sub>NH spectrum of **I**. The VT-<sup>15</sup>N, <sup>13</sup>C labelling was adopted during  $t_1$ , which limits the acquisition time in the <sup>13</sup>C dimension to ca. 5–8 ms due to evolution of carbon–carbon coupling constants. The  $t_1$  acquisition time for <sup>15</sup>N was twice that for <sup>13</sup>C resonances, which was made possible by storing the carbon magnetisation in the form of two-spin order (2C<sub>z</sub>H<sub>z</sub>) during parts of the <sup>15</sup>N labelling (Boelens et al., 1994; Vis et al., 1994).

In principle, chemical shifts of all <sup>13</sup>C resonances, including those of aromatic carbons, can be labelled during the t<sub>1</sub> period of the 3D <sup>13</sup>C, <sup>15</sup>N HSOC-NOESY-CH<sub>3</sub>NH experiment. This would, however, lead to some very long acquisition times since spectral folding is difficult to optimise in this indirectly detected dimension. We have therefore focused on the aliphatic carbons by using an adiabatic band-selective WURST inversion (Kupce and Freeman, 1995). The <sup>13</sup>C inversion pulse starts after the 180° <sup>1</sup>H pulse of the INEPT spin echo and the RF field is swept from aliphatic to aromatic resonances. The length of this pulse, the frequency of the <sup>13</sup>C carrier, the frequency sweep range and the length of the surrounding delays were calculated (Kupče and Freeman, 1997) (see legend of Figure 1) by taking into account the difference in <sup>1</sup>J<sub>CH</sub> in different chemical shift regions (Zwahlen et al., 1997). The calculated length of the INEPT interval (5.2 ms) is, at the same time, optimised for the transfer of magnetisation from NH protons to nitrogen. This part of the sequence can be easily modified should only aromatic or methyl resonances be of interest.

### Conclusions

We have designed a 2D CT-<sup>13</sup>CH<sub>3</sub>, VT-<sup>15</sup>N HSQC experiment which combines constant time <sup>13</sup>C and variable time <sup>15</sup>N chemical shift labelling, and illustrated its usefulness for simultaneous recording of methyl-carbon and nitrogen chemical shifts during the indirectly detected periods of 3D NOESY spectra. The resulting 3D NOESY-CH<sub>3</sub>NH and 3D <sup>13</sup>C,<sup>15</sup>N HSQC-NOESY-CH<sub>3</sub>NH experiments provide the <sup>1</sup>H chemical shifts and the <sup>13</sup>C or <sup>15</sup>N chemical shift, respectively of the directly attached heteronucleus, for protons showing NOE interaction with CH<sub>3</sub> and NH protons. The NOE cross peaks are resolved by the methyl-carbon, or the nitrogen chemical shifts which are acquired with a high digital resolution during a time-shared CT-<sup>13</sup>CH<sub>3</sub>, VT-<sup>15</sup>N interval.

# Acknowledgements

The Edinburgh Centre for Protein Technology is funded by the UK Biotechnology and Biotechnological Research Council (BBSRC) and Department of Trade and Industry and also a grant from the BBSRC Integration of Cellular Responses Initiative to P.N.B. and S.Y.W.

#### References

- Bax, A. and Grzesiek, S. (1993) Acc. Chem. Res., 26, 131–138.
- Boelens, R., Burgering, M., Fogh, R.H. and Kaptein, R. (1994) J.
- Biomol. NMR, 4, 201–213. Clore, G.M., Kay, L.E., Bax, A. and Gronenborn, A.M. (1991)
- Biochemistry, **30**, 12–18. Dayie, K.T. and Wagner, G. (1994) J. Magn. Reson., **A111**, 121– 126.
- Diercks, T., Schwaiger, M. and Kessler, H. (1998) J. Magn. Reson., 130, 335–340.
- Diercks, T., Coles, M. and Kessler, H. (1999) J. Biomol. NMR, 15, 177–180.
- Frenkiel, T., Bauer, C., Carr, M.D., Birdsall, B. and Feeney, J. (1990) J. Magn. Reson., 90, 420–425.

- Gardner, K.H., Rosen, M.K. and Kay, L.E. (1997) *Biochemistry*, 36, 1389–1401.
- Hu, W. and Zuiderweg, E.R.P. (1996) J. Magn. Reson., B113, 70–75.
- Jerala, R. and Rule, G.S. (1995) J. Magn. Reson., B108, 294-298.
- Kay, L.E., Clore, G.M., Bax, A. and Gronenborn, A.M. (1990) Science, 249, 411–414.
- Kupče, E. and Freeman, R. (1995) J. Magn. Reson., A117, 246-256.
- Kupče, E. and Freeman, R. (1997) J. Magn. Reson., 127, 36-48.
- Pascal, S.M., Muhandiram, D.R., Yamazaki, T., Forman-Kay, J.D. and Kay, L.E. (1994) J. Magn. Reson., B103, 197–201.
- Smith, M.A. and Shaka, A.J., 41<sup>th</sup> ENC, Asilomar (2000) Abstract 34, p. 128.
- Sørensen, O.W. (1990) J. Magn. Reson., 89, 210-216.
- Uhrín, D., Uhrínová, S., Leadbeater, C., Nairn, J., Price, N.C. and Barlow, P.N. (2000) J. Magn. Reson., 142, 288–293.
- Vis, H., Boelens, R., Mariani, M., Stroop, R., Vorgias, C.E., Wilson, K.S. and Kaptein, R. (1994) *Biochemistry*, **33**, 14858–14870.
- Zuiderweg, E.R.P., Petros, E.M., Fesik, S.W. and Olejniczak, E.T. (1991) J. Am. Chem. Soc., **113**, 370–372.
- Zwahlen, C., Legault, P., Vincent, S.J.F., Greenblatt, J., Konrat, R. and Kay, L.E. (1997) J. Am. Chem. Soc., **119**, 6711–6721.
- Zwahlen, C., Vincent, S.J.F., Gardner, K.H. and Kay, L.E (1998a) J. Am. Chem. Soc., **120**, 4825–4831.
- Zwahlen, C., Gardner, K.H., Sarma, S.P., Horita, D.A., Byrd, R.A. and Kay, L.E. (1998b) *J. Am. Chem. Soc.*, **120**, 7617–7625.